PREPARATION OF N-TOSYLMETHYLIMINO COMPOUNDS AND THEIR USE
IN THE SYNTHESIS OF OXAZOLES, IMIDAZOLES AND PYRROLES

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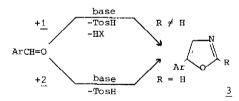
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(Received in UK 14 November 1975; accepted for publication 27 November 1975)

We herein wish to describe the preparation and synthetic applications of a series of N-tosylmethylimino compounds of type 1

Recently, it was shown in our laboratory that tosylmethyl isocyanide (TosMIC, $\underline{2}$) can be used successfully to cycloadd a CH-N=CH fragment to unsaturated substrates 1 This process, involving an addition-cyclisation of TosMIC-anion (TosCHN=C) and concomitant loss of Tos $^{\Theta}$, for example converts aldehydes to oxazoles 2 Thus, the isocyano carbon of $\underline{2}$ ends up in the, by necessity, <u>unsubstituted</u> 2-position of oxazole $\underline{3}$ (R = H) By using imines of type $\underline{1}$, instead of TosMIC ($\underline{2}$), 2-substituted oxazoles $\underline{3}$ (R \neq H) are expected in a related process



We here illustrate the synthetic potentialities of N-tosylmethylimino derivates (1) in the synthesis of a number of oxazoles 3, imidazoles 4 and pyrroles 5 in a single operation at room temperature from aldehydes, aldimines and Michael acceptors, respectively (Scheme and Table I) So far, especially the results obtained with methyl N-tosylmethylthiobenzimidate (1c) seem promising In particular, 1c gives good yields of 2-phenyl substituted pyrroles with Michael acceptors as appears from the examples 5b,e,f

SCHEME and TABLE 3 I

Tosch₂N=C
$$\stackrel{R}{\sim}$$
 $\stackrel{1}{\sim}$ + Arch=chy

+ Arch=nAr'

Ar $\stackrel{N}{\sim}$ R

Ar $\stackrel{N}{\sim}$ R

 $\stackrel{Ar}{\sim}$ $\stackrel{Ar}{\sim}$ R

 $\stackrel{Ar}{\sim}$ $\stackrel{Ar}{$

Product ^a	R	х		Ar	Ar'/Y	yıeld	(%) mp (°C)	base/solvent
	from	ToscH	N=CRX	and alde	hyde			
<u>3a</u>	Me	OMe	(<u>1a</u>)	Ph	-	49	57-59 (rep ⁴ 58-60)	NaH/DME-DMSO
b	Me	оме	(<u>1a</u>)	p-ClPh	-	63	74-75 5	NaH/DME-DMSO
<u>c</u>	Ph	SMe	(<u>1c</u>)	p-0 ₂ NPh	_	50	194 5-195 5	t-BuOK/DME
				-			(rep ⁵ 189 5-190	5)
				and aldı	mine			
<u>4a</u>	Ph	SMe	(<u>1c</u>)	Ph	Ph	23	251-252 5 (subl) (rep ⁶ 248-250)	NaH/DME-DMSO
<u>b</u>	Ph	SMe	(<u>1c</u>)	p-ClPh	p-ClPh	51	203.5-204	Nah/DME-DMSO
<u>c</u>	SMe	SMe	(<u>1d</u>)	p-ClPh	p-ClPh	64	148-149 5	t-BuOK/t-BuOH-DME
				and Mich	ael acceptor			
<u>5a</u>	Me	SMe	(<u>1b</u>)	Ph	COPh	91 ^b	235-236 (sl dec) (rep ⁷ 231)	NaH/DME-DMSO
<u>b</u>	Ph	SMe	(<u>1c</u>)	Ph	COPh	73	199 5-200 5	NaH/DME-DMSO
<u>c</u>	SMe	sMe	(<u>1d</u>)	Ph	COPh	65	171-173 (sl dec)	NaH/DME
₫	OMe	OMe	(<u>1g</u>)	Ph	COPh	42	dec at <u>ca</u> . 100	t-BuOK/THF
<u>e</u>	Ph	SMe	(<u>1c</u>)	Ph	COOMe	58	164 5-165 5	Nah/DME-DMSO
<u>f</u>	Ph	SMe	(<u>1c</u>)	Ph	C≅N	63	266-267 (sl dec)	NaH/DME-DMSO

- a No heterocyclic products were obtained in reactions of p-chlorobenzaldehyde with

 1e or 1c, 1-methylindole-3-carbaldehyde with 1a, N-p-nitrobenzylidenemethylamine with

 1c, cinnamonitrile with 1a and 3-penten-2-one with 1a, reaction of 1e with p-nitro
 benzoylchloride (t-BuOK/DME) yields 2-chloro-5-p-nitrophenyl-4-tosyloxazole (20%),

 mp 187-189° (dimorphous)
- b. The same product was obtained in 75% by using 1a instead of 1b.

To rationalize the formation of the products $\underline{3}$, $\underline{4}$ and $\underline{5}$ it seems logical to assume as the first step the generation of a 2-azaallyl anion (TosCH-N-CRX, $\underline{6}$) by proton abstraction from the activated methylene. ^{8,9} Reaction of $\underline{6}$ with the unsaturated substrate ($\underline{1}$ \underline{e} aldehyde, aldimine or Michael acceptor) could be either a [3+2] 1,3-anionic cycloaddition, ¹⁰ or a [3+2] 1,3-dipolar cycloaddition (after loss of Tos $^{\Theta}$ or x^{Θ}) ¹¹

We have employed the following reactions in the synthesis of the previously unknown N-tosylmethylimino derivates $\underline{1}$, which are fairly stable crystalline solids. They can well be kept at -20° (under N₂). The known N-tosylmethylacetamide ($\underline{7}$), obtained by a Mannich condensation of TosH, CH₂O and CH₃CONH₂, 12 was O-methylated smoothly with methyl fluorosulfonate (Magic Methyl) 13 to give methyl N-tosylmethylacetimidate ($\underline{1a}$, Scheme and Table II). The analogous reaction of N-tosylmethylbenzamide 12 (8) was unsuccessful

Both $\underline{7}$ and $\underline{8}$ were readily converted with P_4S_{10} in DME¹⁴ to the corresponding thioamides $\underline{9}$ (mp 142-143°, 81%) and $\underline{10}$ (mp 142 5-144°, 97%), 15 respectively. Reaction with methyl fluorosulfonate gave in good yields the S-methylated thioimidates $\underline{1b}$ and $\underline{1c}$. A Mannich condensation carried out with methyl dithiocarbamate provided methyl N-tosylmethyldithiocarbamate 16 ($\underline{11}$, mp 150-152°, 72%), from which 1d was obtained in 93% by methylation analogous to $\underline{1b}$, \underline{c}

SCHEME and TABLE II

Compd	R	х	yıeld (%)	mp (°C)
<u>1a</u>	Me	ОМе	80	90-93 (dec) ^a
<u>1b</u>	Ме	SMe	73	103-104 5
<u>1c</u>	Ph	SMe	74	95-97
<u>1d</u>	SMe	SMe	93	122-123
<u>1e</u>	Cl	Cl	68	70-73 5 (rep ¹⁹ 70-73.5)
<u>1 f</u>	MeO	Cl	72	131 5-133
<u>1g</u>	MeO	MeO	65	108-110 (sl dec)

a isolated as HSO₃F salt

Addition of chlorine to TosMIC $(\underline{2})$ yields N-tosylmethyldichloroformimide 18 $(\underline{1e})$ from which either one or both chlorines can be displaced with MeONa in MeOH to give $\underline{1f}$ and $\underline{1g}$, respectively (Table II)

Further synthetic applications of N-tosylmethylimino derivatives ($\underline{1}$) are currently under investigation

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